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Exam Code: 107402 Subject Code: 7033

B.Sc.(Bio Technology) - 2nd Semester (Old sylb. 2019-20) (2721)

Paper: BT-3 Inorganic Chemistry-B

Time allowed: 2 Max. Marks: 40

Note: required to attempt any FOUR questions. There are EIGHT questions of equal marks. Candidates are

Section - A

- (a) N2 and CO are isoeled CO complexes. Offer a thoric but M-N2 complexes are much weaker than Mble explanation. (3)
- (b) Give two methods to nature of bonding linear M-CO group in metal carbonyls. paration of metal carbonyls. Also discuss the (7)
- (a) Draw the stru s of following metal carbonyls: 5
- (b) Write a short note on metal carbonyl hydrides.

(ii) Ir₄(CO)₁₂

(iii) Fe₃(CO)₁₂

(iv) Rh4(CO)12

(5)

Section - B

(a) Define podands. Give two examples.

3 3

- affecting the selectivity of crown ethers (b) Discuss two methods to prepare crown ethers. Also discuss the factors
- (a) Define cryptands. Give two examples. Also give two methods to prepare

S

advantages cryptands (b) Briefly discuss mechanism of phase transfer catalysis. Also give S

Contd....P/2

Section - C

- S (a) Derive relationship between stepwise and commulative stability constants.
- (b) The complexes containing chelate rings are more stable than those containing similar but unidentate ligands. Why is it so? Explain with the 5. 5

help of suitable example

- (a) What do you understand by stability of complex ion? On what factors does the stability of a complex ion depend? Explain giving suitable examples.
- (b) What are porphyrins? Also draw the structure of porphyrin

sodium and potassium ions in the biological

7

(b) Draw photosynthesis. the structure of chlorophyll. Also discuss its important role in

(5)

(5)

- (a) Illustrate the structure of myoglobin and hemoglobin. Discuss in detail the roles played by these bioinorganic compounds in biological systems 3
- (b) Briefly describe the role of carbonic anhydrase in the biological systems.

Co

3

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